

Lab #: 235488 Job #: 17407

Sample Name/Number: HW02z

Company: TechLaw, Inc.
Date Sampled: 1/25/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Water -----

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Chemical  $\delta^{13}C$  $\delta^{18}O$  $\delta D$ Component mol. % % % % Carbon Monoxide ----nd Hydrogen Sulfide ----na Helium -----0.0112 Hydrogen ----nd Argon -----0.628 Oxygen -----0.80 Nitrogen -----40.72 Carbon Dioxide -----0.094 Methane ----- 57.06 -160.6-29.30 Ethane -----0.687 Ethylene ----nd Propane ----nd Propylene -----0.0001 Iso-butane ----nd N-butane ----nd Iso-pentane ----nd N-pentane ----nd Hexanes + ----nd

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 590 Specific gravity, calculated: 0.736

-64.6

-9.66

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.



Lab #: 235489 Job #: 17407

Sample Name/Number: HW04

Company: TechLaw, Inc.
Date Sampled: 1/24/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Component	Chemical	$\delta^{13}$ C	δD	$\delta^{18}O$
component	mol. %	%	‰	‰
Carbon Monoxide	nd			
Hydrogen Sulfide	na			
Helium	na			
Hydrogen	nd			
Argon	1.50			
Oxygen	2.28			
Nitrogen	84.37			
Carbon Dioxide	2.01			
Methane	9.76	-24.98	-121.8	
Ethane	0.0796			
Ethylene	nd			
Propane	0.0004			
Propylene	nd			
Iso-butane	nd			
N-butane	nd			
Iso-pentane	nd			
N-pentane	nd			
Hexanes +	nd			
Water			-63.2	-9.48

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 100 Specific gravity, calculated: 0.947

Remarks:

Analysis is of gas extracted from water by headspace equilibration. Analysis has been corrected for helium added to create headspace. Helium dilution factor = 0.74

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.

<sup>\*</sup>Addition of helium negates the ability to detect native helium and may negate the ability to detect hydrogen.



Lab #: 235490 Job #: 17407

Sample Name/Number: HW05

Company: TechLaw, Inc.
Date Sampled: 1/26/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Component	Chemical	$\delta^{13}$ C	δD	$\delta^{18} O$
Component	mol. %	%	%。	%。
Carbon Monoxide	nd	*		
Hydrogen Sulfide	na			
Helium	na			
Hydrogen	nd			
Argon	1.54			
Oxygen	4.82			
Nitrogen	84.97			
Carbon Dioxide	0.40			
Methane	8.24	-33.0	-162.9	
Ethane	0.0259			
Ethylene	nd			
Propane	nd			
Propylene	nd			
Iso-butane	nd			
N-butane	nd			
Iso-pentane	nd			
N-pentane	nd			
Hexanes +	nd			
Water			-63.2	-9.36

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 84 Specific gravity, calculated: 0.948 Remarks:

Analysis is of gas extracted from water by headspace equilibration. Analysis has been corrected for helium

added to create headspace. Helium dilution factor = 0.68
\*Addition of helium negates the ability to detect native helium and may negate the ability to detect hydrogen.

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.

<sup>\*\*</sup> Mathana aarban instance abtained anline via CC C IDMS

<sup>\*\*</sup> Methane carbon isotopes obtained online via GC-C-IRMS



235491 Job #: Lab #: 17407

Sample Name/Number: HW06

Company: TechLaw, Inc. Date Sampled: 1/26/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Water -----

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Chemical  $\delta^{13}C$  $\delta^{18}O$  $\delta D$ Component mol. % % % % Carbon Monoxide ----nd Hydrogen Sulfide ----na Helium -----0.0248 Hydrogen -----0.0222 Argon -----0.503 Oxygen -----1.04 Nitrogen -----32.03 Carbon Dioxide -----0.008 Methane ----- 65.62 -31.07 -169.0Ethane -----0.746 Ethylene ----nd Propane -----0.0068 Propylene -----0.0001 Iso-butane ----nd N-butane ----nd Iso-pentane ----nd N-pentane ----nd Hexanes + ----nd -65.6 -9.85

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 678 Specific gravity, calculated: 0.700

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.



Lab #: 235492 Job #: 17407

Sample Name/Number: HW08a

Company: TechLaw, Inc.
Date Sampled: 1/25/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Component	Chemical	$\delta^{13}C$	δD	$\delta^{18}O$
component	mol. %	%.	‰	%.
Carbon Monoxide	nd			
Hydrogen Sulfide	na			
Helium	na			
Hydrogen	nd			
Argon	0.746			
Oxygen	5.31			
Nitrogen	36.31			
Carbon Dioxide	3.22			
Methane	53.64	-36.58	-209.9	
Ethane	0.767			
Ethylene	nd			
Propane	0.0030			
Propylene	nd			
Iso-butane	nd			
N-butane	nd			
Iso-pentane	nd			
N-pentane	nd			
Hexanes +	nd			
Water			-61.0	-9.20

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 557 Specific gravity, calculated: 0.774

Remarks:

Analysis is of gas extracted from water by headspace equilibration. Analysis has been corrected for helium added to create headspace. Helium dilution factor = 0.67

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.

<sup>\*</sup>Addition of helium negates the ability to detect native helium and may negate the ability to detect hydrogen.



Lab #: 235493 Job #: 17407

Sample Name/Number: HW12

Company: TechLaw, Inc.
Date Sampled: 1/26/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Chemical  $\delta^{13}C$  $\delta^{18}O$  $\delta D$ Component mol. % % % % Carbon Monoxide ----nd Hydrogen Sulfide ----na Helium -----0.0434 Hydrogen ----nd Argon -----0.115 0.16 Oxygen -----Nitrogen -----4.54 Carbon Dioxide -----0.073 Methane ----- 94.06 -35.90 -196.7Ethane -----0.987 Ethylene ----nd Propane -----0.0221 Propylene -----0.0002 Iso-butane -----0.0006 N-butane -----0.0012 Iso-pentane ----nd N-pentane ----nd Hexanes + ----nd -64.6 -9.60 Water -----

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 972 Specific gravity, calculated: 0.580

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.



Lab #: 235494 Job #: 17407

Sample Name/Number: HW17

Company: TechLaw, Inc.

Date Sampled: 1/27/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Component	Chemical	$\delta^{13}C$	δD	$\delta^{18}O$
Compension	mol. %	%。	‰	%。
Carbon Monoxide	nd			
Hydrogen Sulfide	na			
Helium	na			
Hydrogen	nd			
Argon	1.49			
Oxygen	2.06			
Nitrogen	80.93			
Carbon Dioxide	0.43			
Methane	14.97	-31.54	-167.8	
Ethane	0.118			
Ethylene	nd			
Propane	0.0011			
Propylene	nd			
Iso-butane	nd			
N-butane	nd			
Iso-pentane	nd			
N-pentane	nd			
Hexanes +	nd			
Water			-64.9	-9.63

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 154 Specific gravity, calculated: 0.917

Remarks:

Analysis is of gas extracted from water by headspace equilibration. Analysis has been corrected for helium added to create headspace. Helium dilution factor = 0.72

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.

<sup>\*</sup>Addition of helium negates the ability to detect native helium and may negate the ability to detect hydrogen.



Lab #: 235495 Job #: 17407

Sample Name/Number: HW24

Company: TechLaw, Inc.
Date Sampled: 1/27/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Component	Chemical	$\delta^{13}$ C	δD	$\delta^{18}O$
component	mol. %	%	%。	‰
Carbon Monoxide	nd			
Hydrogen Sulfide	na			
Helium	na			
Hydrogen	nd			
Argon	1.58			
Oxygen	1.29			
Nitrogen	94.00			
Carbon Dioxide	0.017			
Methane	3.11	-53.8	-165	
Ethane	nd			
Ethylene	nd			
Propane	nd			
Propylene	nd			
Iso-butane	nd			
N-butane	nd			
Iso-pentane	nd			
N-pentane	nd			
Hexanes +	nd			
Water			-64.8	-9.70

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 31 Specific gravity, calculated: 0.963

Remarks:

Analysis is of gas extracted from water by headspace equilibration. Analysis has been corrected for helium added to create headspace. Helium dilution factor = 0.65

\*Addition of helium negates the ability to detect native helium and may negate the ability to detect hydrogen.

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.

<sup>\*\*</sup> Methane isotopes obtained online via the GC-C-IRMS



Lab #: 235496 Job #: 17407

Sample Name/Number: HW02

Company: TechLaw, Inc.

Date Sampled: 1/25/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Chemical  $\delta^{13}C$  $\delta^{18}O$  $\delta D$ Component mol. % % % % Carbon Monoxide ----nd Hydrogen Sulfide ----na Helium -----0.0110 Hydrogen ----nd Argon -----0.636 Oxygen -----1.12 Nitrogen -----41.09 Carbon Dioxide -----0.10 Methane ----- 56.36 -160.5-29.36 Ethane -----0.683 Ethylene ----nd Propane ----nd Propylene -----0.0001 Iso-butane ----nd N-butane ----nd Iso-pentane ----nd N-pentane ----nd Hexanes + ----nd -64.5 -9.76Water -----

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 583 Specific gravity, calculated: 0.739

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.



Lab #: 235497 Job #: 17407

Sample Name/Number: HW01

Company: TechLaw, Inc.
Date Sampled: 1/25/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Water -----

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Chemical  $\delta^{13}C$  $\delta^{18}O$  $\delta D$ Component mol. % % % % Carbon Monoxide ----nd Hydrogen Sulfide ----na Helium -----0.0747 Hydrogen ----nd Argon -----0.683 Oxygen -----0.20 Nitrogen -----49.91 Carbon Dioxide -----0.005 Methane ----- 48.69 -202.4-36.80 Ethane -----0.432 Ethylene ----nd Propane -----0.0004 Propylene -----0.0001 Iso-butane ----nd N-butane ----nd Iso-pentane ----nd N-pentane ----nd Hexanes + ----nd

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 501 Specific gravity, calculated: 0.769

-65.1

-9.81

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.



Lab #: 235498 Job #: 17407

Sample Name/Number: HW14

Company: TechLaw, Inc.
Date Sampled: 1/26/2012

Container: Dissolved Gas Bottle

Field/Site Name: A3TA

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Component	Chemical	$\delta^{13}$ C	δD	$\delta^{18}O$
Compension	mol. %	%	%。	%
Carbon Monoxide	nd			
Hydrogen Sulfide	na			
Helium	na			
Hydrogen	nd			
Argon	1.46			
Oxygen	2.70			
Nitrogen	72.02			
Carbon Dioxide	4.99			
Methane	18.74	-26.58	-140.3	
Ethane	0.0899			
Ethylene	nd			
Propane	nd			
Propylene	nd			
Iso-butane	nd			
N-butane	nd			
Iso-pentane	nd			
N-pentane	nd			
Hexanes +	nd			
Water			-63.2	-9.54

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 191 S

Specific gravity, calculated: 0.927

Remarks:

Analysis is of gas extracted from water by headspace equilibration. Analysis has been corrected for helium added to create headspace. Helium dilution factor = 0.74

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.

<sup>\*</sup>Addition of helium negates the ability to detect native helium and may negate the ability to detect hydrogen.



Lab #: 235499 Job #: 17407

Sample Name/Number: HW19

Company: TechLaw, Inc. Date Sampled: 1/23/2012

Dissolved Gas Bottle Container:

Field/Site Name: A3TA

Location:

Formation/Depth: Sampling Point:

Date Received: 2/03/2012 Date Reported: 2/20/2012

Component	Chemical mol. %	δ <sup>13</sup> C ‰	δD ‰	δ <sup>18</sup> Ο ‰
Carbon Monoxide	nd			
Hydrogen Sulfide	na			
Helium	na			
Hydrogen	nd			
Argon	1.63			
Oxygen	7.11			
Nitrogen	86.88			
Carbon Dioxide	4.38			
Methane	0.0011			
Ethane	nd			
Ethylene	nd			
Propane	nd			
Propylene	nd			
Iso-butane	nd			
N-butane	nd			
Iso-pentane	nd			
N-pentane	nd			
Hexanes +	nd			
Water			-61.1	-9.13

Total BTU/cu.ft. dry @ 60deg F & 14.7psia, calculated: 0 Specific gravity, calculated: 1.008

Remarks:

Analysis is of gas extracted from water by headspace equilibration. Analysis has been corrected for helium added to create headspace. Helium dilution factor = 0.73

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.

<sup>\*</sup>Addition of helium negates the ability to detect native helium and may negate the ability to detect hydrogen.